Syllabus of UNDERGRADUATE DEGREE COURSE

Petrochemical Engineering



Rajasthan Technical University, Kota Effective from session: 2018 – 2019

SYLLABUS

2nd Year - IV Semester: B.Tech. (Petrochemical Engineering)

4PC2-01: Introduction to Petrochemicals

Credit: 3 3L+0T+0P Max. Marks: 150 (IA:30, ETE:120) End Term Exam: 3 Hours

SN	Contents	Hours
1	Introduction: Objective, scope and outcome of the course.	1
2	Introduction: Introduction to Petrochemicals, Applications of Petrochemicals, History and growth of petrochemical industry in India, Petrochemical Industries in India.	5
3	Classification Petrochemicals : Based on processing stage, Number of carbon atoms; Petrochemicals from C1, C2, C3, C4, C5, Syngas and Aromatics.	6
4	Petrochemical Feed Stoke: Classification: Solid, Liquid, Gaseous.	4
5	Chemistry and Technology: Production of Methanol, Formaldehyde, Vinyl Chloride, Ethylene oxide , Isopropanol, Cumene, Acrylonitrile, Isoprene, Ethylene glycol, Linear alkyl benzene, Amines.	24
	Total	40

SYLLABUS

2nd Year - IV Semester: B.Tech. (Petrochemical Engineering)

4PC1-03/3PC1-03: Managerial Economics and Financial Accounting

Credit: 2 2L+0T+0P

Max. Marks: 100 (IA:20, ETE:80) End Term Exam: 2 Hours

SN	Contents	Hours
1	Introduction: Objective, scope and outcome of the course.	1
2	Basic economic concepts- Meaning, nature and scope of economics, deductive vs inductive methods, static and dynamics, Economic problems: scarcity and choice, circular flow of economic activity, national income-concepts and measurement.	3
3	Demand and Supply analysis- Demand-types of demand, determinants of demand, demand function, elasticity of demand, demand forecasting –purpose, determinants and methods, Supply-determinants of supply, supply function, elasticity of supply.	5
4	Production and Cost analysis- Theory of production- production function, law of variable proportions, laws of returns to scale, production optimization, least cost combination of inputs, isoquants. Cost concepts-explicit and implicit cost, fixed and variable cost, opportunity cost, sunk costs, cost function, cost curves, cost and output decisions, cost estimation.	5
5	Market structure and pricing theory - Perfect competition, Monopoly, Monopolistic competition, Oligopoly	4
6	Financial statement analysis - Balance sheet and related concepts, profit and loss statement and related concepts, financial ratio analysis, cash-flow analysis, funds- flow analysis, comparative financial statement, analysis and interpretation of financial statements, capital budgeting techniques.	8
	Total	26



SYLLABUS

2nd Year - IV Semester: B.Tech. (Petrochemical Engineering)

4PC1-02/3PC1-02: Technical Communication

Credit: 2 Max. Marks: 100(IA:20, ETH 2L+0T+0P End Term Exam: 2 H		•
SN	Contents	Hours
1	Introduction: Objective, scope and outcome of the course.	1
2	Introduction to Technical Communication- Definition of technical communication, Aspects of technical communication, forms of technical communication, importance of technical communication, technical communication skills (Listening, speaking, writing, reading writing), linguistic ability, style in technical communication.	3
3	Comprehension of Technical Materials/Texts and Information Design & development- Reading of technical texts, Readingand comprehending instructions and technical manuals, Interpreting and summarizing technical texts, Note-making. Introduction of different kinds of technical documents, Information collection, factors affecting information and document design, Strategies for organization, Information design and writing for print and online media.	6
4	Technical Writing, Grammar and Editing - Technical writing process, forms of technical discourse, Writing, drafts and revising, Basics of grammar, common error in writing and speaking, Study of advanced grammar, Editing strategies to achieve appropriate technical style, Introduction to advanced technical communication. Planning, drafting and writing Official Notes, Letters, E-mail, Resume, Job Application, Minutes of Meetings.	8
5	Advanced Technical Writing - Technical Reports, types of technical reports, Characteristics and formats and structure of technical reports. Technical Project Proposals, types of technical proposals, Characteristics and formats and structure of technical proposals. Technical Articles, types of technical articles, Writing strategies, structure and formats of technical articles.	8
	Total	26



SYLLABUS

2nd Year - IV Semester: B.Tech. (Petrochemical Engineering)

4PC3-04: Material Science and Technology (Common with Chemical Engineering)

Credit: 3 3L+0T+0P

Max. Marks: 150 (IA:30, ETE:120) End Term Exam: 3 Hours

SN	Contents	Hours
1	Introduction: Objective, scope and outcome of the course.	1
2	Introduction to materials : Atomic structure, bonding aggregates of atom. Crystals Structure : crystal structure, periodicity in crystal, types of structures: SC, BCC, FCC and HCP Crystals system, crystal lattice, unit cell, crystal direction, crystal planes, Miller indices, inter planar spacing, X-ray analysis, Crystals Defects : classifications and impact on the properties of engineering materials	7
3	Phase Equilibria – phase rule phase changes in pure Iron, binary systems, solid solution, Eutectic, Eutectoid, Peritectic and Peritectoid reactions. General principles of heat treatment: Annealing, normalizing, hardening, tempering and age hardening	8
4	Corrosion: Types of Corrosion in Industries, corrosion of materials in construction, pipe line and in equipments and its control	4
5	Materials and their properties: Mechanical properties: Hardness, Strength, Toughness, Stiffness, Ductility, Malleability, Hardenability, Creep fatigue and Rheology. Electrical properties: Conductors, Semiconductors and insulators, dielectric materials. Optical properties: Absorption, Reflection, Transmission and Refraction, optical fibers and lasers. Magnetic properties: various types of magnetic materials, Diamagnetic, Paramagnetic, Ferromagnetic, Ant ferromagnetic and Ferromagnetic materials, Domain theory, Hard and soft magnetic materials. Thermal Properties: Thermal expansion, Heat capacity, Thermal Conduction, Thermal Stresses. Criteria for selection of materials for special applications in Industries such as smart materials. Smart materials.	10
6	Characterization of Material: Principle, Construction and Procedure for characterization of material using Scanning Electron Microscopy (SEM), Transmission Electron Microscopy (TEM), EDS/EDX, Atomic force microscopy (AFM), Dielectric spectroscopy, Fluorescence spectroscopy.	10
	Total	40

SYLLABUS

2nd Year - IV Semester: B.Tech. (Petrochemical Engineering)

4PC4-05: Heat Transfer (Common with Chemical Engineering)

Credit: 3 3L+0T+0P

Max. Marks: 150 (IA:30, ETE:120) End Term Exam: 3 Hours

SN	Contents	Hours
1	Introduction: Objective, scope and outcome of the course.	1
2	Conduction: Heat transfer modes, laws; General heat equation; Steady state problems in plane and composite systems; Thermal resistance; Insulation and critical radius; unsteady state heat conduction; Extended surfaces as Fins.	9
3	Convection: Principle Heat balance Equation in laminar flow; Natural convectionheat transfer from plate and cylinder.Principles, Dimensional analysis of Heat Transfer by Forced, Principles, Dimensional analysis of Heat Transfer by Natural, Laminar and Turbulent Boundary layers; Laminar and turbulent flow heat transfer in a circular pipe. Dimensional groups in Heat Transfer	6
4	Condensation and Boiling: Types of condensation: Drop and Film condensation, Condensation on a vertical plate, vertical tube and horizontal tubes. Effect of superheated vapor and non-condensable gases. Types of boiling: Pool and forced boiling; boiling curves; Simplified relations for boiling heat transfer with water; Critical Flux.	6
5	Radiation: Basic concepts; Emission characteristics and laws of black body radiation; Radiation incident on a surface; Solid angle and radiation intensity.Heat exchange by radiation between two black surface elements; Heat exchange by radiation between two finite black surfaces; shape factor; Radiation shields.	8
6	Heat Exchangers Classification of heat exchangers; Overall heat transfer coefficient, fouling factor calculations; Analysis of Heat Exchangers: Logarithmic Mean temperature difference, Effectiveness – NTU Method.	6
7	Evaporator: Evaporators, types of evaporator method of feedingsteam consumption, economy.	4
	Total	40



SYLLABUS

2nd Year - IV Semester: B.Tech. (Petrochemical Engineering)

4PC4-06: Mass Transfer-I (Common with Chemical Engineering)

Credit: 3 3L+0T+0P

Max. Marks: 150 (IA:30, ETE:120) End Term Exam: 3 Hours

SN	Contents	Hours
1	Introduction: Objective, scope and outcome of the course.	1
2	Fundamentals of Mass Transfer: Individual and film coefficients, overall mass transfer Coefficient and their inter relationships; Analogies in transfer processes, determination of mass transfer coefficient.	5
3	Diffusion Phenomenon : Molecular and eddy diffusion in gases, liquids and solids, Interface mass transfer. Mass transfer theories: film theory Penetration theory and surface renewal theory.	6
4	Humidification and Dehumidification : Humidification: General Theory, psychometric chart. Fundamental concepts in humidification & dehumidification, wet bulb temperature. Adiabatic saturation temperature, measurement of humidification calculation of humidification operation, cooling towers and related equipments.	10
5	Drying: Equilibrium mechanism theory of drying, drying rate curve. Batch and continuous drying for tray driers, Drum dryers, spray and tunnel dryers.	8
6	Absorption : Introduction to Adsorption, Absorption and Extraction in continuous contact columns; co-current, counter current and cross current contacting Absorption, calculations of NTU and HTU, Concept of HETP, Two phase flow in packed beds, co-current and counter current Processes Flooding loading, column internals: types of trays/ plates and packing, point and plate efficiency.	10
		40



SYLLABUS

2nd Year - IV Semester: B.Tech. (Petrochemical Engineering)

4PC4-07: Applied Thermodynamics (Common with Petroleum Engineering & Chemical Engineering)

Credit: 3
3L+0T+0P

Max. Marks: 150 (IA:30, ETE:120) End Term Exam: 3 Hours

SN	Contents	Hours
1	Introduction: Objective, scope and outcome of the course.	1
2	Review of first and second law of thermodynamics.	7
3	Vapor-liquid equilibrium: phase rule, simple models for VLE;VLE by modified Raoult'slaw;VLE from K-value correlations; Flash calculations.	10
4	Solution Thermodynamics: fundamental property relationships, free energy and chemical potential, partial properties, definition of fugacity and fugacity coefficient of pure species and species in solution, the ideal solution and excess properties. Liquid phase properties from VLE, Models for excess Gibbs energy, heat effects and property change on mixing. UNIFAC and UNIQUAC models. Liquid-Liquid Equilibria; Vapor-Liquid-Liquid Equilibria; Solid- Liquid Equilibria; Solid-GasEquilibria.	12
5	Chemical reaction equilibria: equilibrium criterion, equilibrium constant, evaluation of equilibrium constant at different temperatures, equilibrium conversion of single reactions, multi-reaction equilibria Introduction to molecular/statistical thermodynamics.	10
	1	40

SYLLABUS

2nd Year - IV Semester: B.Tech. (Petrochemical Engineering)

4PC4-21: Heat Transfer Lab

Credit: 2 0L+0T+4P Max. Marks: 100 (IA:60, ETE:40)

1.	To determine the thermal conductivity of Liquid.
2.	To determine the equivalent thermal conductivity of composite wall.
3.	To determine heat transfer coefficient in force convection and natural convection
4.	Study of Unsteady state Heat Transfer Unit
5.	To determine heat transfer coefficient with the help of Stefan Boltzmann Apparatus.
6.	To calculate emissivity of the test plate by emissivity measurement apparatus.
7.	To determine heat transfer coefficient in double pipe heat exchanger.
8.	To study the heat transfer characteristics of a shell and tube heat exchanger.
9.	To measure determine the heat transfer coefficient and heat transfer rate of film wise and drop wise condensation of pure water vapor.
10	. To determine rate of evaporation through single effect evaporator.

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4PC4-22: Mass Transfer Lab

Credit: 2 OL+OT+4P

List of Experiment

Max. Marks: 100 (IA:60, ETE:40)

1.	To determine diffusion coefficient of solid vapour in air
2.	To determine diffusion coefficient of Liquid vapour in air
3.	To study the rate dissolution of a rotating cylinder and then to calculate the
	mass transfer coefficient. (Mass Transfer with and without chemical
	Reaction)
4.	To investigate the mass transfer characteristic of a wetted surface column
	unit.

- 5. To investigate the characteristics of cooling tower.
- 6. To study the drying characteristics of a wet granular material using natural and forced circulation in tray dryer.
- 7. To prepare the drying rate curve for force draft tray dryer.
- 8. To study the characteristics of spray dryer.
- 9. To study Absorption of gas in absorption column

SYLLABUS 2nd Year - IV Semester: B.Tech. (Petrochemical Engineering)

4PC4-23: Comprehensive Analysis of Petrochemical Engineering

Credit: 2 0L+0T+4P Max. Marks: 100 (IA:60, ETE:40)



4PC8-00: Social Outreach, Discipline & Extra Curricular Activities(SODECA)

Credit: 0.5

Max. Marks: 25